

Nanomaterials for Water Purification Towards Global Water Crisis Sustainable Solutions

Akin Zor¹, Alim Rahman²

^{1,2}College of Dentistry, Taibah University, Al Madina Al Munawara 41311, Saudi Arabia

KEYWORDS:

Nanomaterials;
Water Purification;
Sustainable Solutions;
Environmental Remediation;
Filtration Technology

ARTICLE HISTORY:

Submitted : 26.04.2025
Revised : 20.05.2025
Accepted : 23.06.2025

<https://doi.org/10.31838/INES/03.02.02>

ABSTRACT

Providing the world's increasingly growing population with clean, affordable water is an unprecedented challenge for the 21st century. In the face of climate change and failing water quality, nanotechnology has become the center piece in innovation in this global crisis. This article discusses the ways in which nanomaterials are leading the way in the revolution of the water purification techniques and are providing sustainable means to increase access to safe drinking water around the world. Water scarcity and pollution threaten human health, economic development and environmental sustainability. Traditional water treatment methods do not generally achieve effective removal of emerging pollutants and pathogen. Thus, the properties of nanomaterials at the molecular scale make it unique, and open new possibilities for development of more efficient, cost effective and environmentally friendly water purification. Nanotechnology is changing the way we clean and reuse water resources from nano adsorbents that remove toxic metals selectively and nanomembranes that filter out the slightest contaminants. In this article we will look at the advances in nanomaterials for water purification looking at how they work, what they are good for, and they could help enlighten us about the global water crisis. Through these innovations we'll explore ways these are redefining the future of water management for anyone and everyone, not just industries and communities, but all across the globe.

Author e-mail: itsadeelnaz@hotmail.com

How to cite this article: Zor A, Rahman A. Nanomaterials for Water Purification Towards Global Water Crisis Sustainable Solutions. Innovative Reviews in Engineering and Science, Vol. 3, No. 2, 2026 (pp. 13-22).

NANOTECHNOLOGY: WATER PURIFICATION: A GAME CHANGER

Water purification has emerged as a complex challenge that nanotechnology has emerged as a powerful tool in being able to address. Scientists and engineers have discovered extraordinary properties when they manipulate materials at the nanoscale; that is, between 1 and 100 nanometers - and these revolutionary properties can be brought to bear on water treatment processes. Nanomaterials have unique features, such as high surface area to volume ratio, enhanced reactivity as well as a flexibility to be precisely engineered giving them the potential to be applied to a broad class of contaminants in water. Nanomaterials also have potential to enhance water quality, including in the removal from heavy metals and organic pollutants to pathogens and emerging contaminants of concern. Nanotechnology's potential for high efficiency treatment with decreased energy consumption and chemical use makes it one of the

key advantages of nanotechnology in water purification. Indeed, with climate change and resource scarcity on the rise, this is exactly the way to futureproof water management practices.

In addition, nanotechnology affords the use of multifunctional materials that resolve multiple water quality problems at once. As an example, a nanocomposite can include adsorption and antimicrobial properties as well as catalysing properties in one material, providing one such solution to comprehensive water treatment. With the advancement of research in this area, nanotechnology holds promise for additional means of improving clean water access for all, more quickly, cheaply and environmentally. The following sections will focus on various nanomaterials and the applications for their use in water purification, and will discuss how they may revolutionize the way that water is globally purified.^[1-5]

Nanoadsorbents: Precision Removal of Contaminants

The cutting edge class of materials revolutionizing water purification is nanoadsorbents because they selectively remove contaminants from water. In particular, these nanoscale materials have an exceptionally high surface area to volume ratio, essentially corresponding to many times more active sites for adsorption than are available in conventional adsorbents.

The viabilities and efficiencies of nanoadsorbents are very versatile. • lead, mercury, or arsenic heavy metals • Organic pollutants (e.g., pesticides, pharmaceuticals)

Pesticides represent a critical class of persistent organic pollutants that present major challenges for elimination based on their inherent properties, widespread occurrence, and uses in routine practices by the global community. • (inorganic) ions (e.g. fluoride, nitrate) • Emerging contaminants of concern.ion: Sustainable Solutions for Global Water Crisis

The 21st century faces an unprecedented challenge in providing clean, affordable water for the world's growing population. As climate change intensifies and water quality deteriorates, innovative technologies like nanotechnology have emerged as crucial tools for addressing this global crisis. This article explores how nanomaterials are revolutionizing water purification methods, offering sustainable solutions to expand access to safe drinking water worldwide.

Water scarcity and contamination pose severe threats to human health, economic development, and environmental sustainability.

Nanoadsorbents: Precision Removal of Contaminants

- Heavy metals (e.g., lead, mercury, arsenic)
- Organic pollutants (e.g., pesticides, pharmaceuticals)
- Inorganic ions (e.g., fluoride, nitrate)
- Emerging contaminants of concern

Carbon based material such as Carbon nanotubes (CNT) and graphene oxide (GO) are one of the most promising types of nanoadsorbents.

Additionally, easy separation and regeneration is possible because of their magnetic properties, which further increases their practicality in water treatment applications. Further development of functionalized nanoadsorbents extended their capacities. Researchers have boosted selectivity and affinity of nanomaterials for target pollutants by modifying their surface with specific functional groups. Such an approach has resulted

in 'smart' nanoadsorbents that are able to respond to changes in the environment - e.g. pH or temperature - to optimise their performance. Since this is a rapidly maturing area of research, nanoadsorbents are expected to have a significant impact on solving complicated water purification problems. Because they can efficiently remove a multitude of different pollutants, and they can readily be regenerated and reused, they offer a sustainable way to address the issue of improved water quality on a worldwide basis.^[6-9]

Nanomembranes: Can be considered Advanced Filtration Technologies

At once inexpensive, easy to fabricate and incredibly selective, nanomembranes mark a massive step forward in the history of water filtration technology. Typically less than 100 nanometers thick, these ultra thin membranes exploit the special properties of nanomaterials to derive even better filtration performance than conventional membrane technologies.

The key advantages of nanomembranes include:

1. **Enhanced selectivity:** Specific contaminants can be selectively removed from water by the design of precise pore sizes and surface attributes on nanomembranes, while permitting beneficial minerals to pass through.
2. **Improved flux:** Higher water flux rates thanks to the ultra thin nature of nanomembranes (i.e. more water treated with less time), improves overall system efficiency.
3. **Reduced fouling:** Anti-fouling features are incorporated into many nanomembranes to insure enduring performance over time and decrease the frequency with which cleaning or replacement is required.
4. **Lower energy consumption:** Rapid fabrication of nanomembranes, together with their high permeability, often leads to lower operating pressures and, thereby, lower energy requirements of filtration processes.

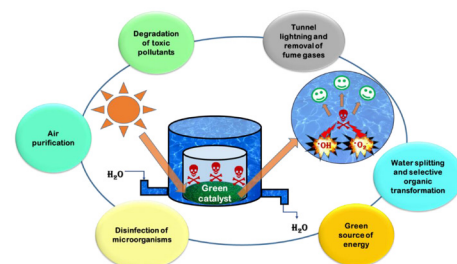


Fig. 1: Nanomembranes: Can be considered Advanced Filtration Technologies.

Several types of nanomembranes have shown promise in water purification applications:

- **Graphene-based membranes:** These ultra thin carbon based membranes have great water permeability and can remove salts, organic molecules and some of the gases from water.
- **Nanocomposite membranes:** These membranes combine the advantages of several materials by incorporating nanoparticles or nanofibers into polymer matrices and enhance both filtration performance and durability.
- **Biomimetic membranes:** Membranes inspired by the natural, biological system are developed with structures and functionalities of biological cell membranes to achieve highly efficient and selective water purification.

Nanomembrane technology is continually being pushed to new material and fabrication limits to solve particular water treatment challenges. For example, the recent advances in mixed matrix nanomembranes have demonstrated promising promise for removing multiple classes of contaminants including: heavy metals, organic pollutants, and pathogens. With continuing development of nanomembrane technology, it may find application in point of use filters, to large scale desalination plants. Nanomembranes' ability to provide water of high quality with little energy and chemical input means they will play a key role in developing sustainable solutions to the global water crisis.^[10-14]

PHOTOCATALYTIC NANOMATERIALS: WATER PURIFICATION USING HARNESSING OF LIGHT

The activation of photocatalytic nanomaterials is one innovative approach to water purification, using light to break down contaminants. At the nanoscale, semiconductors are used as these materials, capable of taking in light energy and causing chemical reactions that break down pollutants into nonpolluting products.

The mechanism of photocatalysis involves several key steps:

1. **Light absorption:** When dealing with ACCEPT doll passes to the final and above state of the instance (final DEFAULT STATE).

The nanomaterial absorbs photons of light with energy greater than or equal to its bandgap.

2. **Electron-hole pair generation:** This absorbed energy makes the material generate electron hole pairs.

3. **Charge separation:** The electrons and holes move up from the nanomaterial and to the surface of the nanomaterial.
4. **Redox reactions:** Although the separated charges themselves do not undergo oxidation or reduction reactions with water and dissolved oxygen, they initiate oxidation and reduction reactions in water and dissolved oxygen to form reactive species (hydroxyl radicals and superoxide ions, i.e.).
5. **Pollutant degradation:** The reactive species attack and break down organic pollutants, pathogen inactivation, and reduces some metal ions.

Currently, one of the most extensively studied photocatalysts is titanium dioxide (TiO₂) nanoparticles due to its relatively high stability, low cost and powerful degrading organic pollutants. However, researchers are continually exploring new materials and strategies to enhance photocatalytic performance:

- **Doped nanomaterials:** One way of extending light absorption of photocatalysts is to incorporate metal or nonmetal dopants into their crystal structure to facilitate charge attachment.
- **Heterojunction nanocomposites:** Charge separation can be improved by combining two or more semiconductors with different band structures, and the range of light absorption is also expanded.
- **Plasmonic photocatalysts:** Surface plasmon resonance effects can enhance light absorption in noble metal nanoparticles integrated to semiconductor photocatalyst.
- **Z-scheme photocatalysts:** Assessing natural photosynthesis these systems mimic natural photosynthesis by using two different photocatalysts in tandem for more efficient use of light energy.

The applications of photocatalytic nanomaterials in water purification are diverse and promising:

- **Organic pollutant degradation:** Persistent organic pollutants-and pesticides, pharmaceutical, and industrial chemicals-can be broken down with photocatalysts.
- **Disinfection:** Strong antimicrobial photocatalysts for effecting inactivation of bacteria, viruses or other pathogens have been demonstrated.
- **Heavy metal reduction:** Photo catalysis could reduce toxic metal ions in the water to less

harmful forms and less harmful forms and even to elemental metals for easy removal.

- **Self-cleaning surfaces:** Adding photocatalysts to membranes or other parts of a water treatment system results in self cleaning surfaces, which maintain performance.

The continued research into this field presents the potential to make photocatalytic nanomaterials an important part of developing sustainable water treatment solutions. Consistent with the mounting recognition of the need for green technologies, their ability to utilize renewable solar energy for such purification processes also works very well.^[15-18]

MAGNETIC NANOPARTICLES: EFFICIENT PRACTICES IN CONTAMINANT REMOVAL AND RECOVERY

Magnetic nanoparticles have proven to be a strong tool in water purification as the ability for contaminant removal and recovery is unlike anything prior. Commonly made up of iron oxides such as magnetite (Fe₃O₄) or maghemite (γ-Fe₂O₃), these nanoscale materials have a high surface area, good reactivity and magnetic properties that make them easy to separate from treated water.

Table 1: Nanomaterials Used for Water Purification

Nanomaterial	Properties
Carbon Nanotubes	Carbon nanotubes have excellent adsorption properties, making them effective for removing organic contaminants and heavy metals from water.
Graphene Oxide	Graphene oxide is known for its high surface area and ability to filter out contaminants, providing a low-cost solution for water purification.
Metal-Organic Frameworks	Metal-organic frameworks offer a highly porous structure that can trap toxic substances, enabling efficient water filtration and removal of pollutants.
Silver Nanoparticles	Silver nanoparticles are effective antimicrobial agents, helping in the removal of bacteria, viruses, and other microorganisms from water sources.
Titanium Dioxide	Titanium dioxide is a photocatalyst that decomposes organic pollutants when exposed to UV light, offering a sustainable solution for purifying water.
Clay Nanocomposites	Clay nanocomposites possess high surface area and adsorption capabilities, making them efficient for filtering out various contaminants in water.

The key benefits of using magnetic nanoparticles in water purification include:

- **High adsorption capacity:** However, such contaminants are easily adsorbed onto their large surface area to volume ratio.
- **Ease of separation:** However, external magnetic fields can separate the nanoparticles quickly and efficiently from treated water.
- **Heavy metal removal:** Functionalized magnetic nanoparticles have been shown to perform well for remediating toxic metals like lead, arsenic and mercury from water.
- **Organic pollutant adsorption:** Magnetic nanocomposites are many which remove organic contaminants, such as dyes, pharmaceuticals, and pesticides.
- **Oil-water separation:** By encapsulating them within hydrophobic magnetic nanoparticles, we could then selectively adsorb oil from oil-water mixtures, which would help to clean up oil spills or improve industrial wastewater treatment.
- **Pathogen removal:** Magnetically responsive nanoparticles with antimicrobial properties, or nanoparticles that can be made functionalized to bind and remove pathogens from water, exist.
- **Nutrient recovery:** Recycling valuable nutrients from wastewater via circular economy approaches in water management can be supported by magnetic nanoparticles used to recover phosphorus (or similar components).

New developments in magnetic nanoparticle technology have included the design and synthesis of stimuli responsive materials which can be bioactivated in reaction to environmental stimuli such as pH or temperature. This gives a much greater degree of control over the capture and release of contaminants on subsequent batches, increasing the overall efficiency and reuse of these adsorbents.

The roles that magnetic nanoparticles will play in developing sustainable water purification solutions as research continues is expected. Efficient contaminant removal capability along with easy separation and potential to be reused, makes them a promising technology for dealing with various water quality challenges at various scales and with different contexts.

- **Improved durability:** However, nanocatalysts are more stable and resistant to deactivation than conventional catalysts are.

Researchers have developed various types of nanocatalysts for water treatment applications:

- **Noble metal nanoparticles:** Nano silver, nano palladium and nano gold have been demonstrated

to have excellent catalytic activity for many types of reactions, including the degradation of organic pollutants and reduction of nitrates.

- **Transition metal oxide nanoparticles:** Oxidation of organic contaminants and removal of heavy metals by catalysts bearing iron, copper, and manganese oxides have been demonstrated.
- **Bimetallic nanocatalysts:** While the catalytic benefits of combining two metals at the nanoscale are well known, the beneficial effect is often limited to specific contaminants.
- **Supported nanocatalysts:** Stability of catalytic nanoparticles can be improved by dispersing them onto supports with high surface area such as activated carbon or zeolites, and the catalyst recovery can be facilitated.

The applications of nanocatalysts in water purification are diverse and promising:

- **Advanced oxidation processes:** Nanocatalyst thus facilitate the generation of reactive oxygen species in Fenton reaction and photocatalytic oxidation processes to provide more efficient degradation of recalcitrant organic pollutants.
- **Reduction of inorganic contaminants:** Nanomaterials have been promising in catalytic

reduction of contaminants such as nitrates, perchlorates and hexavalent chromium.

- **Disinfection:** Nanocatalysts are capable of catalyzing organic synthesis that forms disinfecting agents and some exhibit strong antimicrobial properties.
- **Emerging contaminant removal:** For challenging, emerging contaminants, such as per and polyfluoroalkyl substances (PFAS) and microplastics, nanocatalysts are being developed.

Recent advancements in nanocatalyst technology include the development of:

- **Shape-controlled nanocrystals:** The shape of these catalytic nanoparticles can be precisely controlled to expose specific crystal facets of enhanced activity for particular reactions.
- **Single-atom catalysts:** Expanding the use of individual metal atoms on supports can simultaneously maximize atom efficiency, and achieve unique catalytic properties.
- **Multifunctional nanocomposites:** Catalytic nanoparticles can be combined with other functional materials to form systems that catalyze reactions and adsorb scavenging agents at the same time.

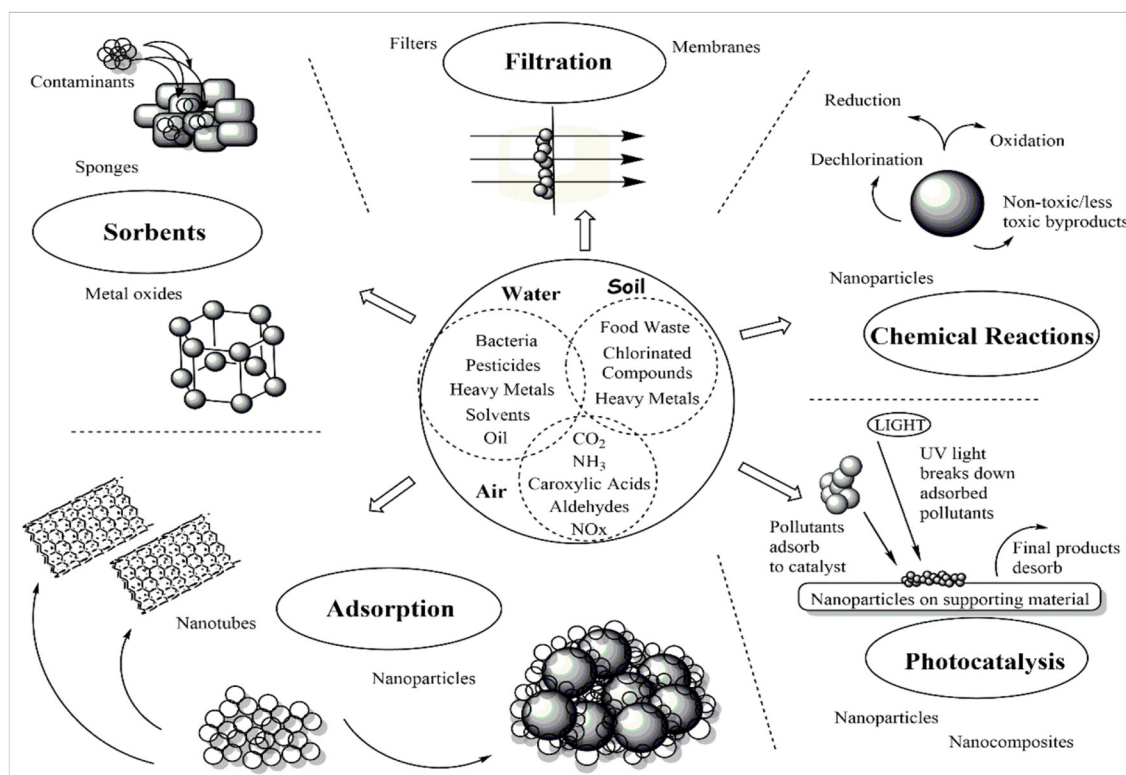


Fig. 2: Magnetic Nanoparticles: Efficient practices in contaminant removal and recovery

Nanocatalysts are prepared to have an important role in the growth of more efficient and ecologically sustainable watersystem technologies with the current research in this area. But their capacity to catalyze reactions at mild conditions and target particular contaminants make them an essential weapon to tackle the many barriers ahead in water purification in the 21st century.

- **Nanobubbles:** Water Treatment Innovative Solutions

Water treatment technologies based on nanobubbles, also known as ultrafine bubbles, are becoming an emerging technology with unique advantages in the purification processes. All of these tiny gas bubbles, less than 100 nanometers in diameter, have extraordinary properties not present in bulk sized bubbles, which open some new doors to water treatment applications.

Key characteristics of nanobubbles include:

- **Long-term stability:** Unlike bigger bubbles, which rise quickly and burst, nanobubbles can stay suspended in water for very long time, perhaps weeks or even months.
- **High internal pressure:** Nanobubbles are formed to be very small in size yet have ultra high internal pressures which makes their reactivity and dissolution rates much more enhanced.
- **Large surface area:** At the nanoscale nanobubbles have a high surface area to volume ratio, thereby aligning to facilitate efficient gas transfer as well as interactions with contaminants.
- **Negative surface charge:** It is known that nanobubbles usually have negative surface charge, and can affect interactions with dissolved species and surfaces.

These unique properties make nanobubbles valuable tools in various water treatment applications:

Oxygenation and aeration: By increasing dissolved oxygen levels in water, oxygen nanobubbles are very efficient at optimizing the processes performed in wastewater treatment and in aquaculture systems.

- **Flotation:** The efficiency of flotation processes, used to separate suspended solids and oils from water, can be improved by the presence of nanobubbles.
- **Disinfection:** Sustained antimicrobial effects in water distribution systems have been provided by nanobubbles containing disinfecting gases such as ozone.
- **Contaminant removal:** As nanobubbles have high surface area and reactivity, they may

contribute to contaminant removal (organic and heavy metals).

- **Membrane cleaning:** As those nanobubbles can clean and maintain the filtration membranes by not fouling and extending the membrane life.

Recent research has explored several innovative applications of nanobubbles in water treatment:

- **Combination with advanced oxidation processes:** Nanobubbles can be integrated with techniques such as UV irradiation or sonolysis for the enhancement of generation of reactive oxygen species for pollutant degradation.
- **Nanobubble-assisted adsorption:** It was found that the presence of nanobubbles can improve the adsorption capacity of various materials, including activated carbon and nanoparticles.
- **Soil and sediment remediation:** Reactive gases or nutrients can be delivered into contaminated soils and sediments using nanobubbles, which can be used as part of an in situ remediation effort.
- **Algae control:** The alterability of water's physicochemical properties from the use of these nanobubbles has shown promise in controlling algal blooms in water bodies.

Now, as research in this field continues, nanobubble technology is poised to have an ever larger role to play in the development of sustainable water treatment solutions. They are good at improving different treatment processes without the need for any added chemicals, which tracks well with the increasing focus on green technologies for solving global water problems.^[15-20]

Hybrid Nanotechnologies: Synergetic Approaches to Water Purification

Hybrid nanotechnologies are a state-of-the-art approach which integrates various nanomaterials or nanotechniques to achieve additive (synergistic) effects. These hybrid systems can take advantage of solving complex water treatment problems better than single component solutions by integrating different nanoscale components.

The key advantages of hybrid nanotechnologies include:

- **Multifunctionality:** Through the combined use of several different nanomaterials it is possible to solve at the same time the treatment of multiple contaminant or water quality problems.
- **Enhanced performance:** Components can work together to enhance efficiency and effectiveness compared with individual technologies.

- **Versatility:** Hybrid systems can be designed to meet particular water treatment requirements in different situations.

RESEARCHERS HAVE DEVELOPED VARIOUS TYPES OF HYBRID NANOTECHNOLOGIES FOR WATER PURIFICATION:

- **Nanocomposite membranes:** By incorporating nanomaterials such as graphene oxide, carbon nanotubes or metal nanoparticles into polymer membranes, the filtration performance can be enhanced, the anti foul properties improved and catalytic or antimicrobial ones added.
- **Magnetic nanocomposites:** Common magnetic nanoparticles are coupled with other functional materials such as activated carbon or graphene oxide to yield systems that combine high adsorption capacity with easy magnetic separation.
- **Photocatalytic hybrids:** Light absorption, charge separation, and total catalytic performance for pollutant degradation can be improved by integrating photocatalysts with other nanomaterials.
- **Nanobubble-assisted technologies:** However, bonding nanobubbles with other remedial procedures, such as adsorption, or advanced oxidation process, can improve total treatment capacity.

Examples of innovative hybrid nanotechnologies in water purification include:

- **Graphene oxide-TiO₂ nanocomposites:** Here we take advantage of adsorption properties of graphene oxide and photocatalytic activity of TiO₂ to create these materials, which provide better removal of both organic and inorganic contaminants.
- **Magnetic core-mesoporous shell nanoparticles:** The magnetic core of these structures facilitates easy separation and the mesoporous shell allows for functionality of the shell to specifically remove contaminants or provide catalytic activities.
- **Nanofibrous membranes with embedded nanoparticles:** Functional nanoparticles incorporated into electrospun nanofibers can provide high filtration efficiency with additional treatment functionalities, such as heavy metal removal or antimicrobial action.
- **Layered double hydroxide-based nanocomposites:** By combining anion exchange, sorption,

and catalytic properties, these materials can serve both to treat water and to catalyze redox reactions.

Future Directions in Nanomaterial Based Water Purification

Nanomaterials possess great potential for revolutionizing water purification, but several challenges have to be overcome in order to fully harness their advantages and adopt them on a widespread, safe, and sustainable basis. A knowledge of these challenges is essential to point future research and development work in this area in the right direction.

Table 2: Advancements in Nanomaterials for Water Purification

Advancement	Development
Enhanced Adsorption Capacity	Enhanced adsorption capacity is achieved by modifying nanomaterials to increase surface area, improving their ability to remove contaminants from water.
Energy-Efficient Filtration	Energy-efficient filtration uses nanomaterials that require minimal energy to filter large volumes of water, making purification processes more sustainable.
Antimicrobial Properties	Antimicrobial properties of certain nanomaterials help eliminate harmful microorganisms, providing safe and clean drinking water in regions with limited resources.
Photocatalysis for Degradation	Photocatalysis for degradation uses light-sensitive nanomaterials, like titanium dioxide, to break down organic pollutants in water when exposed to UV light.
Cost-Effective Synthesis	Cost-effective synthesis of nanomaterials aims to reduce the production cost of water purification technologies, making them accessible for large-scale and developing regions.
Long-Term Stability	Long-term stability is essential for ensuring that nanomaterials used in water filtration can maintain their efficiency over extended periods without degradation.

Key challenges in nanomaterial-based water purification include:

1. **Scalability and cost:** Several nanomaterials exhibit excellent performance in the lab but are limited by scalability to industrial scale applications. Large scale synthesis methods and

the reduction of production costs are critical to the widespread adoption.

2. **Long-term stability and performance:** Durability and availability of nanomaterials to perform consistently over long time spans during real world water treatment conditions is still an issue.
3. **Environmental and health impacts:** Thorough assessment of potential release of nanomaterials into environment and long term effects on the ecosystem and human health is needed and should be mitigated.
4. **Standardization and regulation:** Standards for testing and regulating nanomaterial based water treatment technologies must be pioneered to achieve safety and market adoption.
5. **Integration with existing infrastructure:** Even with much progress, it remains a technical and economics challenge to adapt nanomaterial based solutions to work seamlessly with existing water treatment infrastructure.

Future research directions to address these challenges and advance the field include:

- **Sustainable synthesis methods:** Conceiving green synthesis techniques involving lower energy supply and greener precursors to produce nanomaterial at scale.
- **Smart and responsive nanomaterials:** Genyanthropus blackiSAU-3's brain collage. Credit:© School of Human Evolution and Social Change/ Institute of Human Origins/Arizona State University Graphic by Amy Raub.
- **Hybrid and multifunctional systems:** To advance the design of hybrid nanotechnologies capable of addressing more than one water treatment question and to satisfy diverse treatment needs.
- **In-situ monitoring and control:** Real time monitoring and adaptive control of water этому process through integration of nanosensors and smart materials.
- **Circular economy approaches:** Strategies on how to recover and recycle water treatment nanomaterials to limit waste and environmental impact.
- **Biomimetic and nature-inspired solutions:** Inspiring more efficient and sustainable nanomaterial based systems inspired from natural water purification processes.
- **Machine learning and AI integration:** Advanced data analytics and artificial intelligence leverage

to optimise design and operation of nanomaterial based water treatment systems.

- **Nanomaterial-microorganism interactions:** The synergies between engineered nanomaterials and beneficial microorganisms for improved water treatment performance.

Overcoming these challenges and expanding the boundaries require high output solution research with nanomaterial based water purification boundaries of the global water crisis. Materials scientists, environmental engineers, toxicologists and policymakers will have to collaborate to ensure that these ingenious technologies are safe and workable so that they may be used to help bolster the world's water security.^[21-28]

In confronting the ever increasing global water crisis, the new beacon of hope are nanomaterials, which present innovative and sustainable ways to purify water. Nanomaterials have unique properties that enable them to uniquely address complex water quality challenges with unprecedented efficiency, effectiveness, and they are able to be precisely engineered.

- **Nanoadsorbents** that clean contaminant with unparalleled specificity
- **Nanomembranes** with edge performance beyond the limits of filtration
- **Photocatalytic nanomaterials** used to degrade the pollutants utilizing delivered light energy
- **Magnetic nanoparticles** with contaminant removal and easy separation
- **Mild process acceleration** with nanocatalysts.
- **Nanobubbles** that provide novel solutions for oxygenating and contaminant removal.
- **Hybrid nanotechnologies** that are synergistically multilayer to address multiple water quality problems simultaneously.

- **Nanocomposite membranes:** Incorporating nanomaterials like graphene oxide, carbon nanotubes, or metal nanoparticles into polymer membranes can enhance filtration performance, anti-fouling properties, and even add catalytic or antimicrobial functionalities.
- **Magnetic nanocomposites:** Combining magnetic nanoparticles with other functional materials like activated carbon or graphene oxide creates systems that offer both high adsorption capacity and easy magnetic separation.
- **Photocatalytic hybrids:** Integrating photocatalysts with other nanomaterials can enhance light absorption, charge separation, and overall catalytic performance for pollutant degradation.

Magnetic nanoparticles that combine contaminant removal with easy separation

Nanocatalysts that accelerate treatment processes under mild conditions

- Nanobubbles that offer innovative approaches to oxygenation and contaminant removal
- Hybrid nanotechnologies that synergistically address multiple water quality issues

CONCLUSION

The principles of sustainable development match up well with advancements in nanomaterial based water purification. Nanomaterials provide an avenue for greener and more sustainable water management by allowing for increased use of resources, reduced energy consumption, and decreased chemical inputs. Yet, there is still much to overcome before the full potential of nanomaterials can be fully exploited to solve the global water crisis. But, just as important, the research and development of these technologies must continue towards scaling up production, ensuring long term stability, addressing potential impacts to the environment and human health, and incorporating these technologies into existing infrastructure. The future of nanomaterial water purification is promising. Though these technologies have already arrived, advancing these technologies further can help to enhance their capabilities and sustain their efficiency from the development of smart, responsive materials, to the introduction of artificial intelligence within system design and operation. In doing this, we need to approach the development and implementation of nanomaterial based water purification solutions holistically. That means not only increasing technical capability in these materials but considering other impacts through ecosystems, human health, and socioeconomic systems. Finally, nanomaterials are powerful weapons in the arsenal against the global water crisis. But we can pave the way to a future where clean, safe water is available for everyone by continuing to invest in research, encouraging interdisciplinary collaboration, and encouraging responsible innovation based on what we are discovering about nanomaterials. Nanomaterial based technologies will undoubtedly play a pivotal role as we work to address the water challenges of the 21st century, and beyond, en route to a more sustainable and water secure world.

REFERENCES:

1. Chitpong, N., & Husson, S. M. (2017). Polyacid functionalized cellulose nanofiber membranes for removal of heavy metals from impaired waters. *Journal of Membrane Science*, 523, 418-429.
2. Chung, Y. T., Ba-Abbad, M. M., Mohammad, A. W., & Benamor, A. (2016). Functionalization of zinc oxide (ZnO) nanoparticles and its effects on polysulfone-ZnO membranes. *Desalination and Water Treatment*, 57(17), 7801-7811.
3. Dañbrowski, A. Z. P. E., Hubicki, Z., Podkościelny, P., & Robens, E. (2004). Selective removal of the heavy metal ions from waters and industrial wastewaters by ion-exchange method. *Chemosphere*, 56(2), 91-106.
4. Gómez-Pastora, J., Dominguez, S., Bringas, E., Rivero, M. J., Ortiz, I., & Dionysiou, D. D. (2017). Review and perspectives on the use of magnetic nanophotocatalysts (MNPCs) in water treatment. *Chemical Engineering Journal*, 310, 407-427.
5. Xu Piao, X. P., Zeng GuangMing, Z. G., Huang DanLian, H. D., Feng ChongLing, F. C., Hu ShuAng, H. S., Zhao MeiHua, Z. M., ... & Liu ZhiFeng, L. Z. (2012). Use of iron oxide nanomaterials in wastewater treatment: a review.
6. Cheng, X. L., Xu, Q. Q., Li, S. S., Li, J., Zhou, Y., Zhang, Y., & Li, S. (2021). Oxygen vacancy enhanced Co₃O₄/ZnO nanocomposite with small sized and loose structure for sensitive electroanalysis of Hg (II) in subsidence area water. *Sensors and Actuators B: Chemical*, 326, 128967.
7. GarciaSegura S, Qu X, Alvarez PJJ, Chaplin BP, Chen W, Crit tenden JC, Feng Y, Gao G, He Z, Hou CH, Hu X, Jiang G, Kim JH, Li J, Li Q, Ma J, Ma J, Nienhauser AB, Niu J, Pan B, Quan X, Ronzani F, Villagran D, Waite TD, Walker WS, Wang C, Wong MS, Westerhof P (2020) Opportunities for nanotechnology to enhance electrochemical treatment of pollutants in potable water and industrial wastewater - a perspective. *Environ Sci Nano* 7(8):2178-2194. <https://doi.org/10.1039/D0EN00194E> 121.
8. Gangadhar, C., Moutteyan, M., Vallabhuni, R. R., Vijayan, V. P., Sharma, N., & Theivadas, R. (2023). Analysis of optimization algorithms for stability and convergence for natural language processing using deep learning algorithms. *Measurement: Sensors*, 27, 100784.
9. Garcia-Segura, S., Ocon, J. D., & Chong, M. N. (2018). Electrochemical oxidation remediation of real wastewater effluents—A review. *Process Safety and Environmental Protection*, 113, 48-67.
10. Hasan, I., Khan, R. A., Alharbi, W., Alharbi, K. H., & Alsalme, A. (2019). In situ copolymerized polyacrylamide cellulose supported Fe₃O₄ magnetic nanocomposites for adsorptive removal of Pb (II): artificial neural network modeling and experimental studies. *Nanomaterials*, 9(12), 1687.
11. Herrera-Morales, J., Turley, T. A., Betancourt-Ponce, M., & Nicolau, E. (2019). Nanocellulose-block copolymer films for the removal of emerging organic contaminants from aqueous solutions. *Materials*, 12(2), 230.
12. Chang, J. W., Chen, H. L., Su, H. J., Liao, P. C., Guo, H. R., & Lee, C. C. (2011). Simultaneous exposure of non-diabetics to high levels of dioxins and mercury increases their risk of insulin resistance. *Journal of hazardous materials*, 185(2-3), 749-755.
13. Khan, S. U., Zaidi, R., Shaik, F., Farooqi, I. H., Azam, A., Abuhim, H., & Ahmed, F. (2021). Evaluation of Fe-Mg binary

- oxide for As (III) adsorption—synthesis, characterization and kinetic modelling. *Nanomaterials*, 11(3), 805.
14. Sajab, M. S., Chia, C. H., Chan, C. H., Zakaria, S., Kaco, H., Chook, S. W., & Chin, S. X. (2016). Bifunctional graphene oxide-cellulose nanofibril aerogel loaded with Fe (III) for the removal of cationic dye via simultaneous adsorption and Fenton oxidation. *RSC advances*, 6(24), 19819-19825.
 15. Babu, D. V., Basha, S. A., Kavitha, D., Nisha, A. S. A., Vallabhuni, R. R., & Radha, N. (2023). Digital code modulation-based MIMO system for underwater localization and navigation using MAP algorithm. *Soft Computing*, 1-9.
 16. Fagan, S. B., Santos, E. J. G., Souza Filho, A. G., Mendes Filho, J., & Fazzio, A. (2007). Ab initio study of 2, 3, 7, 8-tetrachlorinated dibenzo-p-dioxin adsorption on single wall carbon nanotubes. *Chemical Physics Letters*, 437(1-3), 79-82.
 17. Huang, Z. H., Zheng, X., Lv, W., Wang, M., Yang, Q. H., & Kang, F. (2011). Adsorption of lead (II) ions from aqueous solution on low-temperature exfoliated graphene nanosheets. *Langmuir*, 27(12), 7558-7562.
 18. Jackson, P., Jacobsen, N. R., Baun, A., Birkedal, R., Kühnel, D., Jensen, K. A., ... & Wallin, H. (2013). Bioaccumulation and ecotoxicity of carbon nanotubes. *Chemistry Central Journal*, 7, 1-21.
 19. Jeon, S., Yun, J., Lee, Y. S., & Kim, H. I. (2010). Removal of Cu (II) ions by Alginate/Carbon Nanotube/Maghemite Composite Magnetic Beads. *Carbon letters*, 11(2), 117-121.
 20. Zhang, X., Lin, M., Lin, X., Zhang, C., Wei, H., Zhang, H., & Yang, B. (2014). Polypyrrole-enveloped Pd and Fe₃O₄ nanoparticle binary hollow and bowl-like superstructures as recyclable catalysts for industrial wastewater treatment. *ACS applied materials & interfaces*, 6(1), 450-458.
 21. Zhang, P., Hou, D., O'Connor, D., Li, X., Pehkonen, S., Varma, R. S., & Wang, X. (2018). Green and size-specific synthesis of stable Fe-Cu oxides as earth-abundant adsorbents for malachite green removal. *ACS Sustainable Chemistry & Engineering*, 6(7), 9229-9236.
 22. Selvam, L., Garg, S., Prasad, R. M., Qamar, S., Lakshmi, K. M., & Ratna, V. R. (2023). Collaborative autonomous system based wireless security in signal processing using deep learning techniques. *Optik*, 272, 170313.
 23. Lee, S. Y., Shim, H. E., Yang, J. E., Choi, Y. J., & Jeon, J. (2019). Continuous flow removal of anionic dyes in water by chitosan-functionalized iron oxide nanoparticles incorporated in a dextran gel column. *Nanomaterials*, 9(8), 1164.
 24. Börjesson, M., Sahlin, K., Bernin, D., & Westman, G. (2018). Increased thermal stability of nanocellulose composites by functionalization of the sulfate groups on cellulose nanocrystals with azetidinium ions. *Journal of Applied Polymer Science*, 135(10), 45963.
 25. Ye, X., Lendel, C., Langton, M., Olsson, R. T., & Hedenqvist, M. S. (2019). Protein nanofibrils: Preparation, properties, and possible applications in industrial nanomaterials. In *Industrial applications of nanomaterials* (pp. 29-63). Elsevier.
 26. Ling, S., Li, C., Adamcik, J., Shao, Z., Chen, X., & Mezzena, R. (2014). Modulating materials by orthogonally oriented β -strands: composites of amyloid and silk fibroin fibrils. *Advanced Materials*, 26(26), 4569-4574.
 27. Hot, J., Topalov, J., Ringot, E., & Bertron, A. (2017). Investigation on parameters affecting the effectiveness of photocatalytic functional coatings to degrade NO: TiO₂ amount on surface, illumination, and substrate roughness. *International Journal of Photoenergy*, 2017(1), 6241615.
 28. Zakaria, R., & Zaki, F. M. (2024). Vehicular ad-hoc networks (VANETs) for enhancing road safety and efficiency. *Progress in Electronics and Communication Engineering*, 2(1), 27-38. <https://doi.org/10.31838/PECE/02.01.03>
 29. Uvarajan, K. P. (2024). Advanced modulation schemes for enhancing data throughput in 5G RF communication networks. *SCCTS Journal of Embedded Systems Design and Applications*, 1(1), 7-12. <https://doi.org/10.31838/ESA/01.01.02>
 30. Velliangiri, A. (2024). Security challenges and solutions in IoT-based wireless sensor networks. *Journal of Wireless Sensor Networks and IoT*, 1(1), 8-14. <https://doi.org/10.31838/WSNIOT/01.01.02>
 31. Kavitha, M. (2024). Enhancing security and privacy in reconfigurable computing: Challenges and methods. *SCCTS Transactions on Reconfigurable Computing*, 1(1), 16-20. <https://doi.org/10.31838/RCC/01.01.04>
 32. Kavitha, M. (2024). Energy-efficient algorithms for machine learning on embedded systems. *Journal of Integrated VLSI, Embedded and Computing Technologies*, 1(1), 16-20. <https://doi.org/10.31838/JIVCT/01.01.04>
 33. Kavitha, M. (2023). Beamforming techniques for optimizing massive MIMO and spatial multiplexing. *National Journal of RF Engineering and Wireless Communication*, 1(1), 30-38. <https://doi.org/10.31838/RFMW/01.01.04>
 34. Zorpette, G., Sengur, A., & Urban, J. E. (2023). Technological improvements in green technology and their consequences. *International Journal of Communication and Computer Technologies*, 11(2), 1-6. <https://doi.org/10.31838/IJCCTS/11.02.01>
 35. Beyene, F., Negash, K., Semeon, G., & Getachew, B. (2023). CMOS Technology: Conventional Module Design for Faster Data Computations. *Journal of VLSI Circuits and Systems*, 5(1), 42-48. <https://doi.org/10.31838/jvcs/05.01.06>
 36. Kigarura, M., Okunki, L., & Nbende, P. (2023). Primary frontiers in designing and benchmarking the applications of helical antennas. *National Journal of Antennas and Propagation*, 5(2), 7-13.